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Energy Levels and Sublevels ...

Considering your answer to question 5,
how many electrons can fit in the entire 5th
energy level? can (A Vel is a ha AC5a)

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50đ 7. Write the notation for an electron spinning clockwise in a p sublevel ...

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WORKSHEET The Activity Series Use
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the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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8 . 1.7, the CF bond is polar covalent. (b)

!ENCl"P = ENCl " ENP. = 3.0" 2.1

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!ENCI"P = 0.9. Since $0.5 < \Delta EN < 1.7$,
the PCl bond is polar covalent. (c)

!ENCI"Li Section 8.5 Electronegativity
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polar . that is in excess.

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